
1st Semester
Branch: Common
Subject Name: Applied Chemistry

Time Allowed : 3 Hrs.

MM: 60

Section -A

Note: Multiple Choice questions. All questions are compulsory.

$$6 \times 1 = 6$$

Section-B

Note: Objective/Completion type questions. All questions are compulsory. $6 \times 1 = 6$

Q.7 Define modern periodic law.

Q.8 The earthy impurities present in the ore are known as-----.

Q.9 Define normality.

Q.10 The full form of LPG is -----.

Q.11 SI unit of viscosity is -----.

Q.12 Define carbon based nano materials.

Section –C

Note: Short answer type Questions. Attempt any eight questions out of ten questions. $8 \times 4 = 32$

Q.13 Write a short note on metallic bond.

Q.14 Define alloys. What are the purpose of making alloys?

Q.15 Define pH. What are industrial applications of pH?

Q.16 Define scale & sludge. Write 2 disadvantages of scale & sludge.

Q.17 What are the characteristics of an ideal fuel?(4 only)

Q.18 Briefly explain the properties and uses of CNG.

Q.19 Write a short note on cementation.

Q.20 Write 4 applications of nano materials.

Q.21 State 4 points of difference between orbit and orbital.

Q.22 Define the following terms---- Ductility, Elasticity

Section-D

Note: Long answer questions. Attempt any two questions out of three questions.
2x8=16

Q.23 Define permanent hardness of water. Discuss the ion exchange method used for the removal of permanent hardness of water.

Q.24 Define the following terms--- oiliness, viscosity, viscosity index, flash point

Q.25 Define polymer. State difference between addition and condensation polymers. (6 points)